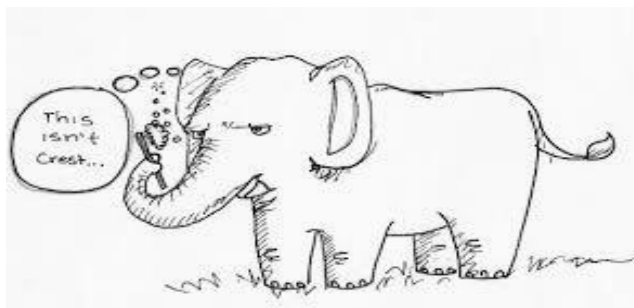


# Elephant's Toothpaste

From Steve Spangler Science

(Because of the volume of foam produced, this is better as a demonstration)



## KEY CONCEPTS/VOCABULARY (What you need to know)

- Chemistry is the science of changing things by mixing them or changing the temperature
- When we mix things, we can cause a chemical reaction
- Some reactions need a catalyst—that one special ingredient that will get the action started
- Yeast is a catalyst
- Catalyst: something that starts or speeds up a chemical reaction
- Exothermic: a chemical reaction that produces heat
- observation: something you saw, heard or noticed
- hypothesis: an educated guess as to what will happen in an experiment

## QUESTION (What you want to learn and what to ask the children)

*These questions should be asked in order so that children learn that a hypothesis is a guess about what will happen in the experiment. This expands their science vocabulary.*

- What will happen if we add food coloring and soap to the jar? What is your hypothesis or guess?
- Will the soap bubble by itself or does it need to be mixed up?
- What will happen if we add yeast? What is your hypothesis or guess?

## EXPERIMENT

### Materials (What you need)

16oz soda bottle, tray, 20 volume hydrogen peroxide (a stronger solution that can be found at a beauty supply store), food coloring, blue Dawn dish detergent, funnel, measuring cup, small container, very warm water, yeast

### Procedures (What to do)

- Tell the children that yeast are a catalyst—that they can cause or speed up a chemical reaction
- Ask the Questions and listen to their hypotheses
- Place soda bottle on tray and using a funnel, add  $\frac{1}{2}$  cup hydrogen peroxide
- Add a couple drops of food coloring and a squirt of Dawn
- In a separate container, dissolve 1 tsp of yeast into very warm water
- Using the funnel, add the yeast to the coke bottle and quickly remove the funnel
- Have the children feel the warm foam that results

## CONCLUSION

### Results (What happened?)

- The yeast causes the peroxide to release oxygen very quickly, forming lots of foam that overflows the bottle
- This reaction is exothermic—both the foam and the bottle will feel warm

### Questions to Ask

- How does the foam feel?